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Key indicators

Single-crystal X-ray study T = 295 K Mean σ (C–C) = 0.003 Å R factor = 0.030 wR factor = 0.085 Data-to-parameter ratio = 16.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

catena-Poly[[(2,2'-biquinoline- $\kappa^2 N, N'$)copper(I)]- μ -isothiocyanato- $\kappa^2 N$:S]

The isothiocyanate ligand in the title compound, $[Cu(NCS)(C_{18}H_{12}N_2)]_n$, links adjacent tetrahedral Cu^I atoms into a zigzag chain; the Cu lies on a mirror plane that bisects the chelating biquinoline ligand.

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Comment

The crystal structures of a number of copper isothiocyanate complexes of 2,2'-bipyridine have been reported, *e.g.* bis(2,2'-bipyridine)(isothiocyanato)copper(II) thiocyanate (Kabesova *et al.* 1993), *catena-*(2,2'-bipyridine)sesqui(μ_2 -thiocyanato)-hemiisocyanatocopper(II) (Ferlay *et al.*, 1999), (2,2'-bipyridine)bis- μ_2 isothiocyanatocopper(II) (Diaz *et al.*, 1999; Madalan *et al.*, 2002; Parker *et al.*, 1994) and *catena-*(5,5'-dimethyl-2,2'-bipyridine)- μ -isothiocyanatocopper(II) (Albada *et al.*, 2004). The 2,2'-biquinoline heterocycle is an analog of 2,2'-bipyridine and, although the synthesis of the compound has been known for a long time (Crowne & Breckenridge, 1954), it is not widely used in the formation of coordination complexes. The only known crystal structure of a copper complex is that of tetrahedral bis(2,2'-biquinoline)copper(I) tetrafluoroborate, which has been reported recently (Ali *et al.*, 2006).



The reaction of this *N*-heterocycle with copper(I) thiocyanate affords a monochelated compound, (I), in which the isothiocyanate functions as a bridging unit (Fig. 1); the mode of bridging gives rise to a zigzag chain that propagates along the *c* axis of the orthorhombic unit cell. The Cu^I center, which lies on a special position of site symmetry *m*, exhibits a distorted tetrahedral coordination. The bite angle of the heterocycle is similar to those found in the 2,2'-bipyridine adducts. The *N*-heterocycle straddles a mirror plane that relates the two quinoline units; the two ring systems are twisted by 10.1 (1)° with respect to each other.

Experimental

© 2006 International Union of Crystallography All rights reserved A mixture of copper(I) thiocyanate (0.121 g, 1.0 mmol), sodium thiocyanate (0.081 g, 1.0 mmol), 2,2'-biquinoline (0.256 g, 1.0 mmol),

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water (5.0 ml) and acetonitrile (5 ml) was sealed in a 25 ml Teflonlined stainless steel Parr bomb. The bomb was heated in an oven to 433 K for 72 h, and was then cooled to room temperature at a rate of 5 K h⁻¹. Brown, almost black, crystals were obtained in about 50% yield. Elemental analysis calculated for $C_{19}H_{12}CuN_3S$: C 60.38, H 3.20, N 11.12%; found C 60.25, H 3.15, N 11.23%.

Z = 4

 $D_x = 1.612 \text{ Mg m}^{-3}$

 $0.16 \times 0.14 \times 0.10 \ \mathrm{mm}$

7944 measured reflections

1844 independent reflections

1606 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation

 $\mu = 1.54 \text{ mm}^{-1}$

T = 295 (2) K

Column black

 $R_{\rm int} = 0.021$

 $\theta_{\rm max} = 27.5^{\circ}$

Crystal data

 $\begin{bmatrix} Cu(NCS)(C_{18}H_{12}N_2) \end{bmatrix} \\ M_r = 377.92 \\ Orthorhombic, Pnma \\ a = 9.0249 (6) Å \\ b = 15.855 (1) Å \\ c = 10.8836 (7) Å \\ V = 1557.4 (2) Å^3 \\ \end{bmatrix}$

Data collection

Bruker APEX area-detector diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\min} = 0.663, T_{\max} = 0.862$

Refinement

 $\begin{array}{ll} \mbox{Refinement on } F^2 & w = 1/[\sigma^2(F_{\rm o}^2) + (0.0486P)^2 \\ R[F^2 > 2\sigma(F^2)] = 0.030 & w \mbox{here } P = (F_{\rm o}^2 + 2F_{\rm c}^2)/3 \\ w R(F^2) = 0.085 & w \mbox{here } P = (F_{\rm o}^2 + 2F_{\rm c}^2)/3 \\ S = 1.07 & (\Delta/\sigma)_{\rm max} = 0.001 \\ 1844 \mbox{ reflections } & \Delta\rho_{\rm max} = 0.41 \mbox{ e } {\rm \AA}^{-3} \\ 115 \mbox{ parameters } & \Delta\rho_{\rm min} = -0.17 \mbox{ e } {\rm \AA}^{-3} \\ \mbox{H-atom parameters constrained } \end{array}$

Table 1

Selected geometric parameters (Å, °).

Cu1-N1	2.107 (1)	Cu1-N2 ⁱⁱ	1.931 (2)
Cu1-N1 ⁱ	2.107 (1)	Cu1-S1	2.286 (1)
N1-Cu1-N1 ⁱ	77.79 (8)	N1 ⁱ -Cu1-N2 ⁱⁱ	112.21 (6)
N1-Cu1-N2 ⁱⁱ	112.21 (6)	$N2^{ii}$ -Cu1-S1	123.28 (7)
N1-Cu1-S1	111.19 (4)	N1 ⁱ -Cu1-S1	111.19 (4)
	. 3 (11) 1	. 3	

Symmetry codes: (i) $x, -y + \frac{3}{2}, z$; (ii) $x - \frac{1}{2}, y, -z + \frac{3}{2}$.

H atoms were placed in calculated positions $[C-H = 0.93 \text{ Å} \text{ and } U_{iso}(H) = 1.2U_{eq}(C)]$ and were included in the refinement in the riding-model approximation.

Data collection: *SMART* (Bruker, 2003); cell refinement: *SAINT* (Bruker, 2003); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics:



Figure 1

Plot illustrating the coordination environment of the Cu atom in (I). Displacement ellipsoids are plotted at the 50% probability level. [Symmetry codes: (i) $x, \frac{3}{2} - y, z$; (ii) $x - \frac{1}{2}, y, \frac{3}{2} - z$.]

ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXL97.

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